

## Preliminary Notes

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### Metabolic interrelationships between soluble and microsomal RNA in rat-liver cytoplasm

In the present communication we wish to report the results of some experiments *in vitro*, suggesting a metabolic transfer of radioactivity from soluble polynucleotides of the cell sap to microsomal ribonucleoprotein particles and *vice versa*, under conditions which are known to favour amino acid incorporation into these particles. Both processes require ATP.

*Transfer of radioactivity from s-RNA to m-RNA.* When pH-5 enzymes (*cf.* HOAGLAND *et al.*<sup>1</sup>) obtained from rat liver labelled *in vivo* with radioactive inorganic phosphate were incubated with a  $15,000 \times g$  liver supernatant for different periods of time in the presence of ATP and PGA, the ribonucleoprotein particles isolated from the microsomes by means of deoxycholate (*vide infra*) become labelled. This labelling is completed in about 5 min. Omission of ATP and PGA from the reaction mixture results in a decreased transfer, whereas the specific activity of microsomal RNA drops even further in the presence of 0.02 M arsenate, showing a definite requirement for ATP.

These preliminary experiments do not establish the polynucleotide nature of the precursor in this reaction process. Essentially identical results were obtained, however, with s-RNA, isolated from pH-5 enzymes by means of zone electrophoresis on starch. As was shown in a previous communication<sup>2</sup>, pH-5 enzymes contain s-RNA as a free polynucleotide which migrates in the electric field as a narrow and distinct zone, leaving behind protein, ATP and AMP.

Inorganic phosphate travels at about the same rate as s-RNA. Following electrophoresis (4.5 V/cm, 16 h, phosphate buffer, pH 7.4, I, 0.025) radioactive s-RNA was eluted from the starch segments and dialysed against 10 l distilled water for about 16 h in order to remove free inorganic phosphate. The dialysed RNA was incubated in varying amounts with  $15,000 \times g$  liver supernatant in the presence of ATP and PGA (see legend of Fig. 1). Microsomes isolated from the incubation mixture by centrifugation at  $105,000 \times g$  for 30 min were solubilized in 0.5 % Na-deoxycholate according to LITTLEFIELD *et al.*<sup>3</sup> and centrifuged for another 90 min at  $105,000 \times g$ . The sedimented particles were extracted with a 10 % NaCl solution at 100° and RNA was precipitated by 3 vol. cold ethanol. RNA was plated and counted in a gas flow counter, the amount being determined by reading absorbance at 260 m $\mu$  following hydrolysis in 5 % HClO<sub>4</sub>.

Fig. 1 shows that RNA isolated from the microsomal particles becomes labelled almost in a linear fashion with increasing concentrations of labelled s-RNA. Very recently HULTIN AND VON DER DECKEN<sup>4</sup> reported a similar metabolic transfer of soluble polynucleotides, which they obtained in a different way (phenol extraction).

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Abbreviations: AMP, GMP, CMP, and UMP, adenylic, guanylic, cytidylic, and uridylic acid; ATP, adenosine triphosphate; PGA, 3-phosphoglyceric acid; s-RNA, soluble RNA; m-RNA, microsomal RNA; Tris, tris(hydroxymethyl)aminomethane.

*Transfer of radioactivity from m-RNA to the soluble fraction.* Liver microsomes were isolated from rats treated previously with radioactive inorganic phosphate, and resuspended in the homogenisation medium described by LITTLEFIELD AND KELLER<sup>5</sup>. The suspension was centrifuged for 2 min at about  $500 \times g$  and the supernatant was incubated with soluble enzymes supplemented with ATP and PGA for different periods of time (see legend Fig. 2). The soluble enzymes were obtained by dialysing a  $105,000 \times g$  liver supernatant for 20 h against 10 l 0.01 M Tris buffer, pH 7.4. Following incubation microsomes were removed by centrifugation at  $105,000 \times g$  for 2 h. 1 mg carrier RNA (from yeast) was added to 5 ml of the supernatant and RNA was isolated in the conventional way<sup>1</sup>.

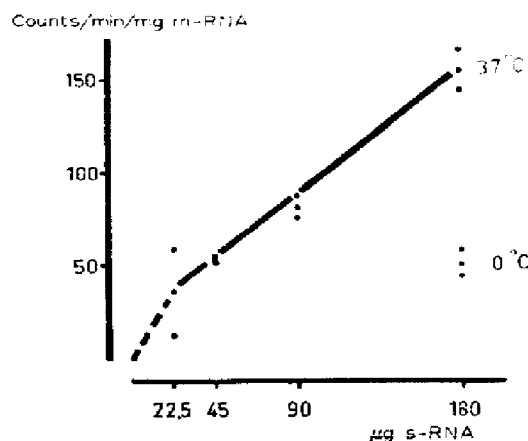


Fig. 1. Transfer of label from [ $^{32}\text{P}$ ]s-RNA to m-RNA. Incubation mixture: 6 ml of  $15,000 \times g$  liver supernatant (non-radioactive), 10  $\mu\text{moles}$  ATP, 80  $\mu\text{moles}$  PGA and [ $^{32}\text{P}$ ]s-RNA in the amounts as indicated. Total vol., 11 ml. Incubation in air for 10 min at 0° or 37°. Each point represents the result of one incubation experiment, the circles (○) indicating the average of two duplicates.

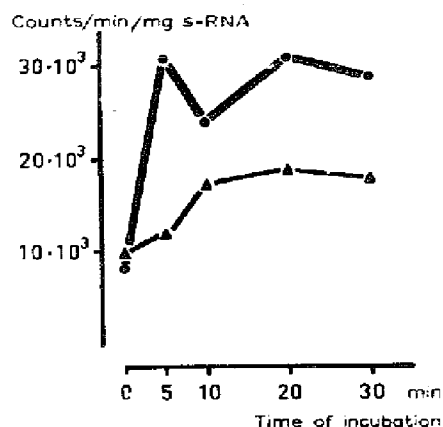


Fig. 2. Transfer of soluble polynucleotides from  $^{32}\text{P}$ -labelled microsomes to the medium. Incubation mixture:  $^{32}\text{P}$ -labelled microsomes (containing 7.2 mg protein, 2.7 mg RNA), dialysed  $105,000 \times g$  liver supernatant (containing 140 mg protein, 0.55 mg RNA), 5  $\mu\text{moles}$  ATP and 40  $\mu\text{moles}$  PGA. Total vol., 5 ml. Incubation in air at 37°. ● In the presence of 5  $\mu\text{moles}$  ATP and 40  $\mu\text{moles}$  PGA. ▲ In the presence of 100  $\mu\text{moles}$  K-arsenate and absence of ATP and PGA.

When the microsomes were incubated in the presence of ATP and PGA the specific activity of s-RNA in the medium increased about 3-fold in 5 min, then remaining constant (Fig. 2). The radioactive s-RNA was non-dialysable and yielded radioactive AMP, GMP, UMP and CMP on alkaline hydrolysis and isolation of the nucleotides on Dowex I (formate). The release of soluble polynucleotides from the microsomes was considerably less when the incubation mixture was supplemented with K-arsenate instead of PGA and ATP. Similar curves were obtained when the medium contained 0.01 M fluoride in addition to 0.02 M arsenate.

Summarizing it may be said that a metabolic relationship seems to exist between soluble and microsomal RNA in rat-liver cytoplasm, which reveals itself by energy-dependent transfers of poly- or oligonucleotides in both directions.

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<sup>1</sup> M. B. HOAGLAND, T. L. STEPHENSON, J. F. SCOTT, L. I. HECHT AND P. C. ZAMECNIK, *J. Biol. Chem.*, 231 (1958) 241.

<sup>2</sup> H. BLOEMENDAL AND L. BOSCH, *Biochim. Biophys. Acta*, in the press.

<sup>3</sup> J. W. LITTLEFIELD, E. B. KELLER, J. GROSS AND P. C. ZAMECNIK, *J. Biol. Chem.*, 217 (1955) 111.

<sup>4</sup> T. HULTIN AND A. VON DER DUEFEN, *Exptl. Cell Research*, 16 (1959) 444.

<sup>5</sup> J. W. LITTLEFIELD AND E. B. KELLER, *J. Biol. Chem.*, 224 (1957) 13.

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### Enzymic cleavage of phosphoramidic acid

The chemical hydrolysis of phosphoramidate has been thoroughly investigated by RATHLEV AND ROSENBERG<sup>1</sup>. Their results indicate that at pH values above 4 the rate of hydrolysis is proportional to the concentration of the anion. Phosphoramidate has been suggested as a metabolic intermediate both by RATHLEV AND ROSENBERG<sup>1</sup> and by SPECK<sup>2</sup>, but no evidence for its participation in metabolic sequences has been presented. In view of its chemical reactivity<sup>3</sup> and the recent evidence for metabolic activity of adenylic-5'-phosphoramidate<sup>4</sup>, a reinvestigation of possible enzymic reactions of phosphoramidate was undertaken.

Sonic extracts prepared from *Escherichia coli* (Crookes strain), grown on a mineral-salts medium with succinate as the sole carbon source, catalyzed a rapid evolution of  $\text{NH}_3$  from phosphoramidate. Boiled extracts were without catalytic activity. Treatment of the *E. coli* extracts with protamine sulfate followed by fractionation with  $(\text{NH}_4)_2\text{SO}_4$  and rigorous dialysis of the fractions revealed at least two separate enzyme systems capable of catalyzing the release of  $\text{NH}_3$  from phosphoramidate. Fraction I, which was precipitated by  $(\text{NH}_4)_2\text{SO}_4$  below 0.5 saturation, was shown to require a divalent metal ( $\text{Mg}^{++}$  or  $\text{Mn}^{++}$ ) and a sulfhydryl compound (cysteine or glutathione) to achieve a maximum rate of  $\text{NH}_3$  release (Table I). The optimum pH for activity of Fraction I was 7.4. Paper chromatography of a reaction mixture containing Fraction I, phosphoramidate, cysteine and  $\text{Mg}^{++}$  with *n*-propanol- $\text{NH}_4\text{OH-H}_2\text{O}$  (6:3:1) as a developing solvent revealed inorganic phosphate and phosphoramidate as the only phosphate-containing compounds present. It is assumed from these results that the reaction catalyzed by Fraction I is the cleavage of phosphoramidate to phosphate and  $\text{NH}_3$ . Even with the mildest conditions phosphate and phosphoramidate cannot be differentiated colorimetrically since molybdate catalyzes a very rapid hydrolysis of the latter<sup>5</sup>.

Another *E. coli* fraction, Fraction III, precipitated between 0.58 and 0.9 saturation with  $(\text{NH}_4)_2\text{SO}_4$ , was also shown to catalyze a rapid evolution of  $\text{NH}_3$  from phosphoramidate, while the intermediate fraction (0.5 to 0.58 saturation) was low in catalytic activity. The rate of  $\text{NH}_3$  release catalyzed by Fraction III was maximal at pH 5.1 and was not stimulated by the addition of either divalent metals or reducing